

**Response Under 37 CFR 1.116**

**Expedited Procedure**

**Examining Group 1753**

Appl. No. 10/522,461

Amdt. dated August 13, 2007

Reply to Office Action of March 14, 2007

Attorney Docket No. 1455-050205

**Amendments to the Specification:**

**Please replace the paragraph beginning at page 6, line 11, with the following amended paragraph:**

-- Thallium atoms have such a simple electronic structure that only three energy states exist below energy of 30,000 cm<sup>-1</sup>, such as the ground state ( $6\ ^2P_{1/2}$ : 0 cm<sup>-1</sup>), metastable state ( $6\ ^2P_{3/2}$ : 7,793 cm<sup>-1</sup>), and the excited state ( $7\ ^2S_{1/2}$ : 26,447.6 cm<sup>-1</sup>). And thallium atoms have several advantages for efficient optical pumping.

These are:

- (a) large electric dipole moment of the transition between  $6\ ^2P_{1/2}$  and  $6\ ^2S_{1/2}$   $7\ ^2S_{1/2}$
- (b) very short level lifetime of  $6\ ^2S_{1/2}$   $7\ ^2S_{1/2}$  (7.5 nsec)
- (c) branching ratio of the transition between  $6\ ^2S_{1/2}$   $7\ ^2S_{1/2}$  and the metastable state is bigger than that of the transition between  $6\ ^2S_{1/2}$   $7\ ^2S_{1/2}$  and the ground state.

Hence, very efficient pumping of thallium atoms into the metastable state can be easily achieved if a CW laser frequency (about 378 nm in the wavelength) is resonant to the transition line of  $6\ ^2P_{1/2}$  and  $6\ ^2S_{1/2}$   $7\ ^2S_{1/2}$ . Because the metastable state population of thallium atoms is lower than  $10^{-3}$  when thallium is heated at temperature to generate an atomic beam, their initial population does not affect the isotope selectivity at this temperature range. --